

# Detached Melt and Vapor Growth of InI in SUBSA Furnace

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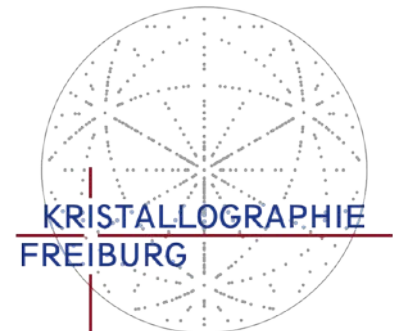
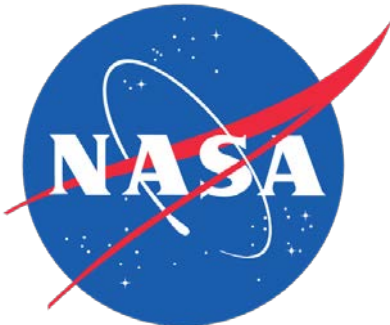
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# OVERVIEW

- **Introduction**
- **SUBSA furnace**
- **Wetting angle results for InI on different substrates**
- **Melt growth and vapor growth ampoule setups**
- **Summary**

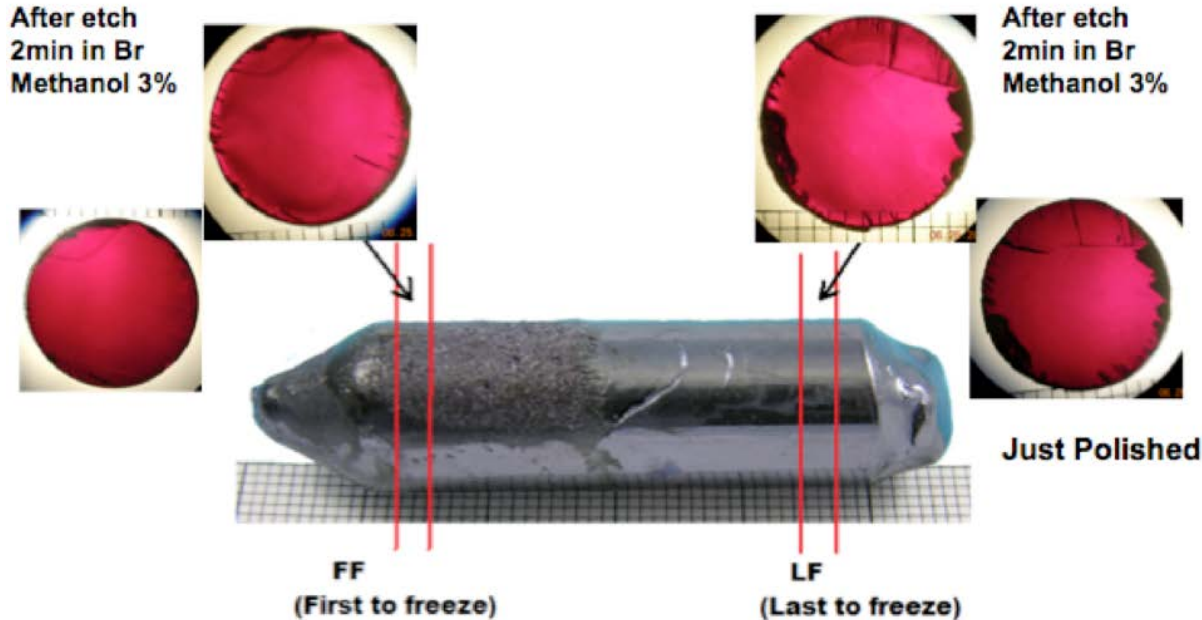
# INI MATERIAL PROPERTIES

- **Monovalent Indium (I) Iodide, InI**, apart from its current uses in metal halide lamps and as reagent in organic synthesis, is a promising candidate for  $\gamma$ -ray and X-ray detectors

Material	$\text{Cd}_{0.9}\text{Zn}_{0.1}\text{Te}$ (CZT)	$\text{HgI}_2$	InI
Average atomic number, Z	49.1	62	51
Density, $\text{g/cm}^3$	5.78	6.4	5.31
Band gap, eV	1.55	2.14	2.0
Melting point, $^{\circ}\text{C}$	$\sim 1100$	259	351
Structure	Zincblende	Tetrahedral-layered	Orthorhombic
Knoop Hardness, $\text{kg/mm}^2$	92	10	27
Molecule Disassoc. Energy eV Herzberg's tables [19]	1.2	0.35	3.43
Electrical Resistivity, Ohm-cm	$3 \times 10^{10}$	$10^{13}$ to $10^{14}$	$5 \times 10^{11}$

# ADVANTAGES OF INI

- The low melting point and congruent sublimation allow both melt growth and vapor growth.
- InI is not toxic and not hygroscopic



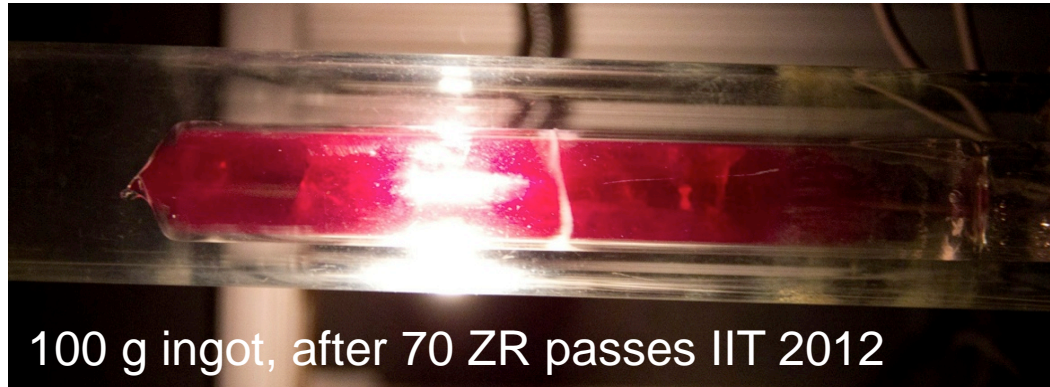
*VB-grown crystal and wafers (IIT)*



*CZ-grown crystals (IIT)*

# PURITY OF THE STARTING MATERIAL

Illinois Institute of Technology and Radiation Monitoring Devices have developed considerable experience in purifying In and I, synthesizing InI from the elements, and purifying InI by zone refining.

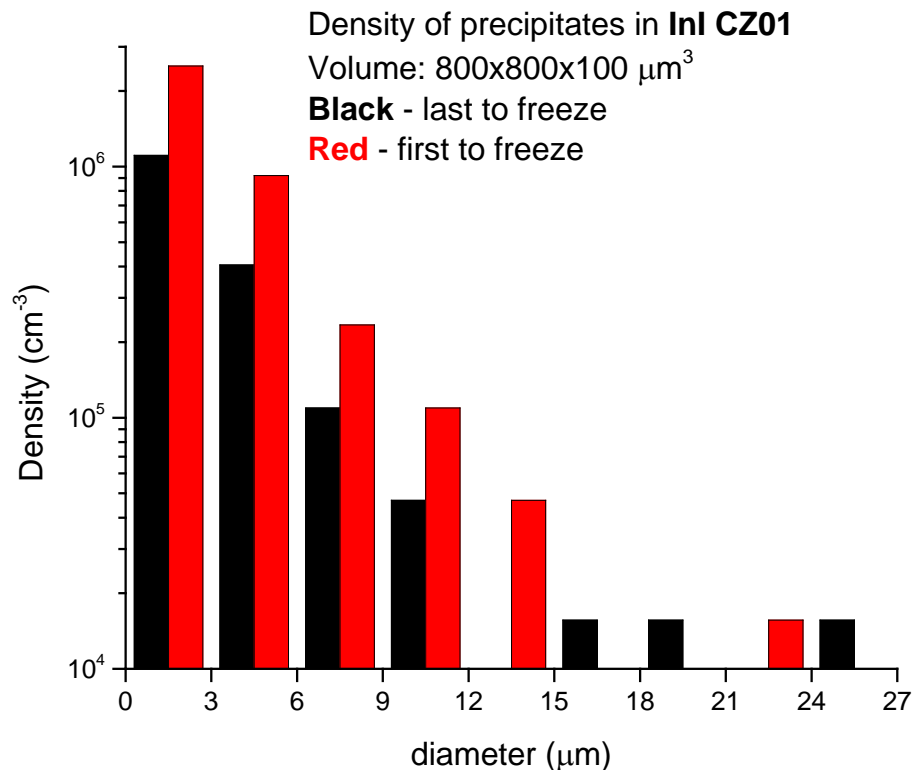


350 g ingot, was ZR and grown in an open boat, under dynamic gas flow 5% H<sub>2</sub> + 95 %Argon, RMD 2015.

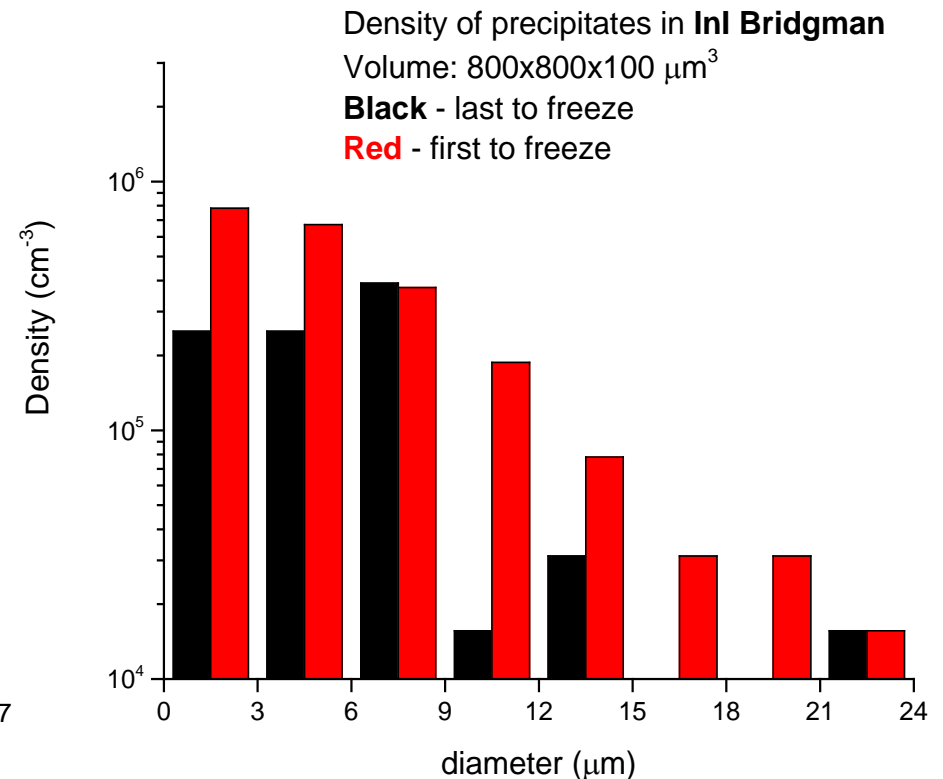
# INCLUSIONS

Small inclusions in the grown material are thought to be responsible for reduced electronic properties, compared to theoretical values

## CZOCHRALSKI



## BRIDGMAN



# GROWTH IN MICROGRAVITY

- Allows vapor growth under purely diffusive conditions, which has shown to lead to a significantly increased  $\mu$ - $\tau$  product in the case of  $\text{HgI}_2$  [1]
- Enhances the chance for detachment in the case of Bridgman growth to reduce stress in the crystal

[1] L. Van den Berg and W.F. Schneppe: Mercuric iodide crystal growth in space. Nucl. Instrum. A 283 (1989), 335-338



# “SUBSA” FURNACE

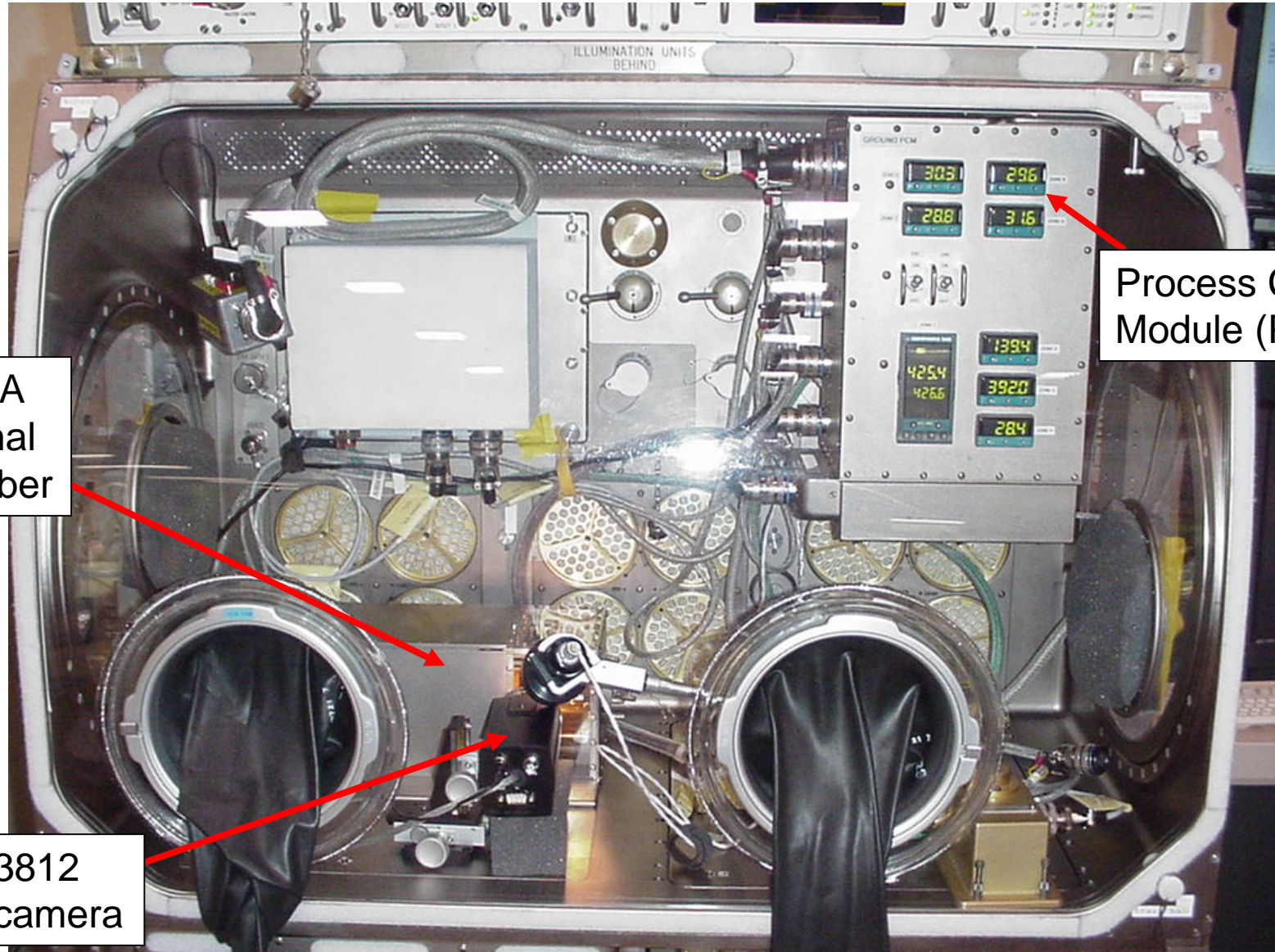
- Developed for a series of InSb experiments with a submerged baffle in the ISS MSG (microgravity science glovebox) rack
- One heating zone, low power consumption,  $T_{\max}=950^{\circ}\text{C}$
- Transparent gradient zone (with ITO radiation shields) to visualize the growth interface



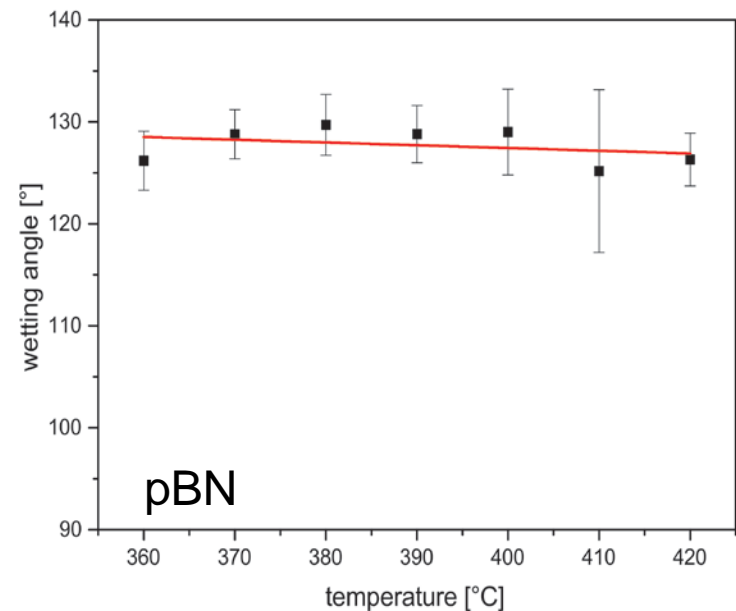
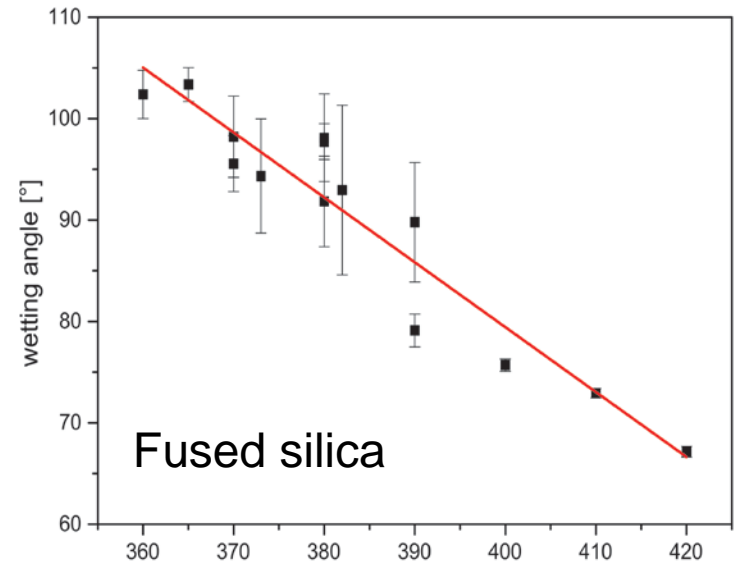
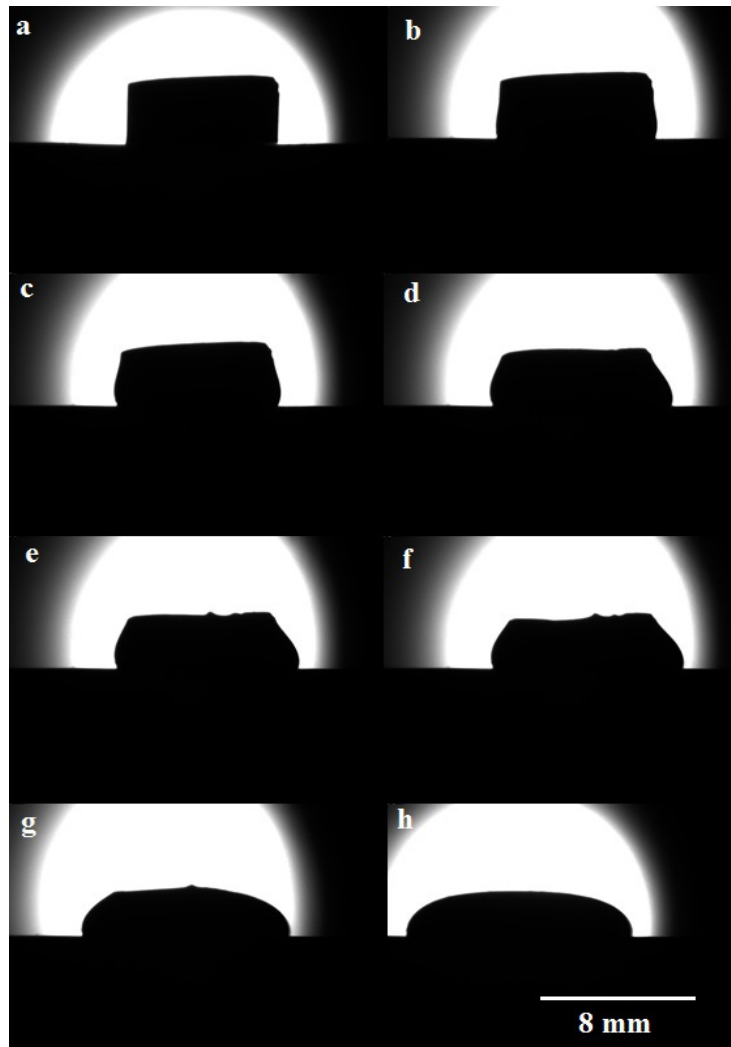
Photo courtesy of Scott Gilley, TecMasters



# SUBSA FURNACE IN THE MSG

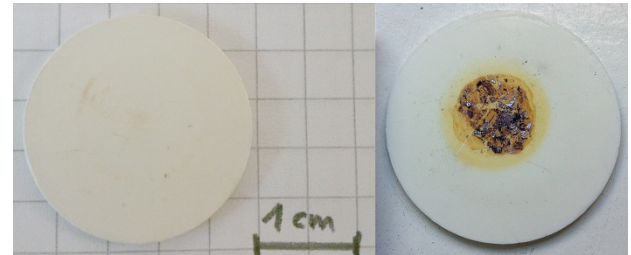


# WETTING: SESSILE DROP TESTS



# WETTING: SESSILE DROP RESULTS

- Wetting angle of InI on fused silica:  $105 \pm 2.6^\circ$   
temperature coefficient:  $-0.64 \pm 0.06^\circ/\text{K}$  (!), sample slides off the substrate easily
- Wetting angle of InI on pBN:  $128 \pm 2.3^\circ$   
temperature coefficient:  $-0.03 \pm 0.03^\circ/\text{K}$ , **but** sample sticks and reacts with the substrate  
( $3\text{InI} + 2\text{BN} \rightleftharpoons \text{In}_3\text{BN}_2 + \text{BI}_3$  ?)



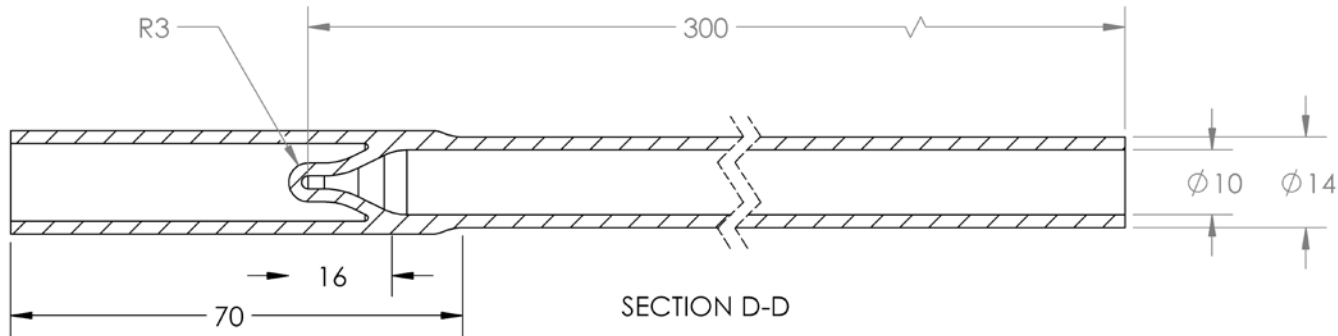
- Wetting angle of InI on  $\text{Al}_2\text{O}_3$  ceramics:  $101.6 \pm 2.1^\circ$   
temperature coefficient:  $-0.315 \pm 0.007^\circ/\text{K}$  [1]
- Wetting angle of InI on carbon:  $93.4 \pm 0.7^\circ$   
temperature coefficient:  $-0.102 \pm 0.006^\circ/\text{K}$  [1]

[1] S.C. Fischer, Ph.D. thesis, Technical University of Aachen and FZ Jülich, 2009. In: Schriften des Forschungszentrums Jülich, Energy & Environment Vol. 44 (2009), pp. 44, 46, 73-75, 135.

<http://hdl.handle.net/2128/3722>

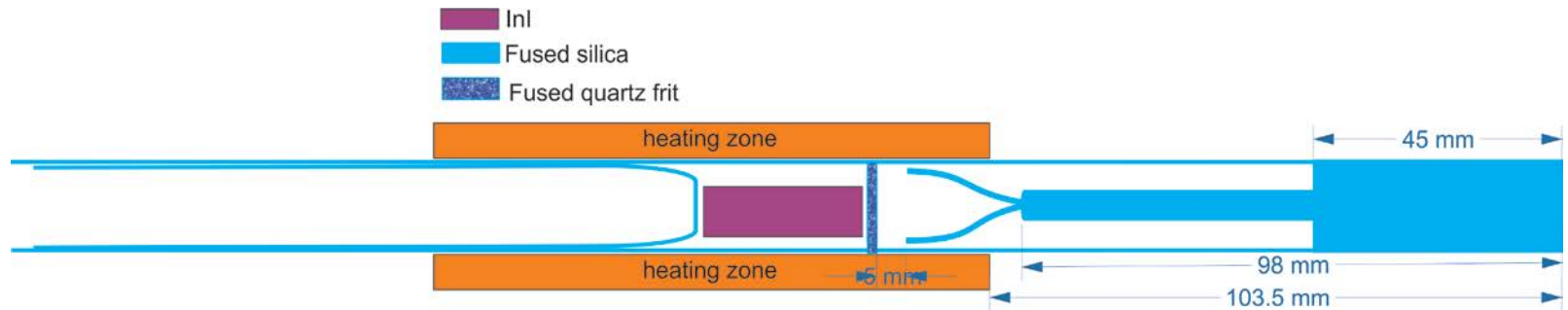


# BRIDGMAN GROWTH AMPOULE



# VAPOR PHASE GROWTH AMPOULE

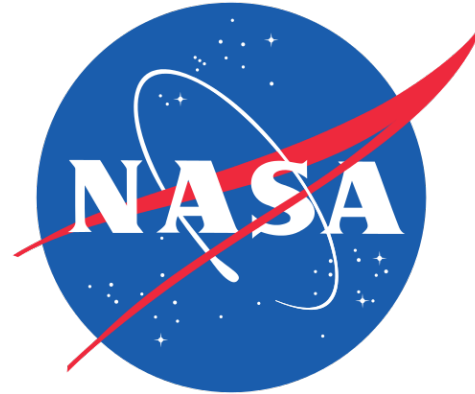
## Modified Markov method setup (semi-closed system)



# SUMMARY

- **InI has promising properties as a room temperature  $\gamma$ -ray detector**
- **The low melting point allows Bridgman growth and CZ growth, although diameter control is difficult with the latter method**
- **Determination of the wetting angles with different crucible materials showed the highest value ( $128^\circ$ ) for pBN, but InI and pBN react, whereas fused silica ( $105^\circ$ ) does not react. Other materials better suited for detachment have to be tested, e.g. DLC coatings.**
- **The congruent sublimation allows vapor phase growth in comparison to melt growth**
- **The melt growth and vapor growth ampoules were launched to the ISS on the Orbital ATK OA-7 Cygnus mission on 18 April, 2017.**
- **Sample processing on the ISS is expected in fall, 2017 or early 2018.**

# ACKNOWLEDGEMENTS



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